Heteroannulation of chromene derivatives. Synthesis of chromeno[4,3-*e*] indazolone, chromeno[4,3-*f*]quinazoline and pyrano[3,2-*c*]chromene derivatives Mahmoud R. Mahmoud*, Fakhry A. El-Bassiouny, Mohamed E. Azab, Mohamed Y. El-Kady and Hesham M. Rashed

Pentane-2,4-dione and 3-ethoxycarbonylcoumarin react in the presence of sodium ethoxide to form 10-acetyl-7,9dihydroxy-6*H*-benzo[*c*]chromen-6-one (2). Compound 2 reacted with aromatic aldehydes and ethyl cyanoacetate to give the chromeno-chromenediones 3 and 4 respectively, with hydrazine hydrate to form the azine 5, with phenyl hydrazine giving the chromeno-indazole 6, with primary amines to form the imines 7a–f, and with thiourea to give the chromeno-quinazoline 8. Methyl 2-amino-4-(4-methoxyphenyl)-5-oxo-4,5-dihydro-pyrano[3,2-*c*]chromene-3-carboxylate (9a) and the corresponding 3-carbonitrile 9b were prepared, and the reactions of the ester 9a with phenacyl chloride, furoyl chloride, and hydrazine hydrate were investigated.

Keywords: 1-benzopyrans, fused pyrans, pyrazoles, indazoles, pyrimidines, coumarins, phenols, imines

Chromene derivatives exhibit various biological activities. They act as anticoagulants, photointercalants or enzyme inhibitors,^{1,2} antibacterials,³⁻⁵ fungicides,⁶ anti-inflammatory⁷ and antitumor^{8,9} agents, and several studies from the viewpoint of chemical taxonomy^{10,11} have been made. A variety of pyrans and condensed pyrans have been prepared using nitriles as starting materials.¹²⁻¹⁵ Within this context and also as a contribution to our work on the synthesis of new heterocyclic derivatives of potential biological activity,¹⁶⁻²⁰ we present here our efforts in the synthesis of several new compounds featuring different heterocyclic rings fused onto the chromene moiety with the aim of obtaining more and better pharmacologically active compounds.

Results and discussion

The reaction of 3-ethoxycarbonylcoumarins with pentane-2,4dione has previously been claimed to afford 1-acetyl-2-methyl-4a,10b-dihydro-4*H*,5*H*-pyrano[3,4 *c*][1]benzopyran-4,5-dione (1) as the major product in a moderate yield.²¹ We find that no pyranobenzopyrone derivative could be isolated in such a reaction; instead we have found that the sole product obtained, in good yield, is the benzo[*c*]chromene derivative **2**. [The precise relationship of the compound reported in ref. 21 as having structure **1** to the product reported here as structure **2** is not clarified. We believe that it might be the same compound; the melting-point reported²¹ for structure **1** (199 °C) is fairly close to what we find for **2** (192–194 °C). It should be noted that the IR bands reported for the C=O groups of **1** (1680 and 1640 cm⁻¹) are not compatible with that structure.]

Structure 1 could be eliminated, since the ¹H NMR spectrum revealed a single 3H singlet at δ 2.4 ppm corresponding to the acetyl group, a broad signal at δ 4.8 ppm (2H), exchangeable with D₂O, characteristic for the two phenolic groups; no signal attributable to the ring methyl group in 1 was observed. Furthermore, the dark green colour of the alcoholic solution formed with neutral ferric chloride indicates the presence

of phenolic OH groups. The remaining spectroscopic and analytical data lent further support to structure 2(see Experimental). The reaction proceeded by a simple Michael addition followed by cyclisation and dehydrogenation to give the more stable product 2.

The reactions of **2** with electrophilic and nucleophilic reagents were also investigated. Thus, treatment of compound **2** with aromatic aldehydes, namely 4-methoxybenzaldehyde, 4-(dimethylamino)benzaldehyde, 4-chlorobenzaldehyde and cinnamaldehyde, in the presence of anhydrous sodium methoxide by fusion at $170 \,^{\circ}$ C or by reflux in n-butanol in the presence of a few drops of piperidine afforded 3-aryl-6-hydroxy-2,3-dihydro-1*H*,7*H*-chromeno[6,5-*c*]chromene-1,7-diones **3a–d**.

The structures of compounds **3** were deduced from their microanalytical and spectral data. The ¹H NMR spectra show the ABX pattern of the CO–CH₂–CH moiety. The magnetic nonequivalence of the protons H_A and H_B of the methylene group adjacent to the carbonyl shows an AB system associated with these protons with a geminal coupling of 11.3 Hz and affords the eight lines of the AB part of an ABX system,²² on account of its attachment to an asymmetric centre.

Fusing compound **2** with ethyl cyanoacetate in the presence of metallic sodium afforded 6-hydroxy-1-methyl-3,7-dioxochromeno[4,3-*f*]chromene-2-carbonitrile (4). (Scheme 2)

The reaction of **2** with hydrazine hydrate (80%) yielded the corresponding azine **5** in which hydrazine has condensed with two molecules of **2** through nucleophilic attack at the ketonic side chain. On the other hand, the reaction of **2** by fusion with phenylhydrazine afforded 5-hydroxy-1-methyl-3-phenylchromeno[4,3-e]indazol-6(3H)-one (**6**). Refluxing of **2** with primary amines, namely aniline, benzylamine, n-butylamine, dodecylamine, ethylamine, hexdecylamine, octdecylamine, and p-toluidine, in n-butanol yielded the corresponding imine condensation products **7a–h** (Scheme 2).



Scheme 1



Reagents: a, ArCHO/NaOMe; b, NCCH₂CO₂Et/Na/xylene; c, N₂H₄.H₂O/n-BuOH; d, PhNHNH₂; e, RNH₂/n-BuOH; f, H₂NCSNH₂/AcOH

Scheme 2

When compound **2** was reacted with thiourea in boiling acetic acid it yielded 6-hydroxy-1-methyl-3-thioxo-3,4-dihydrochromeno[4,3-*f*]quinazolin-7-one (**8**), as evidenced by the microanalytical and spectroscopic data (see Experimental).

characterised 4*H*-Pyrano[3,2-*c*]benzopyrones by а phenanthrene-like structure as found in tetrahydrocannabinol that belongs to the few CNS active compounds without nitogen heteroatoms.23 Heber24 has synthesised socalled azocannabinoids from the reaction of 4-amino-7hydroxycoumarins with α , β -unsaturated carbonyl compounds. In the present work, the reaction of 4-hydroxycoumarin with ethyl-a-cyano-4-methoxycinnamate and a-cyano-4-methoxycinnamonitrile in the presence of anhydrous sodium methoxide in absolute methanol under reflux yielded the corresponding 2-amino-2-carbomethoxy(cyano)-4-(4methoxyphenyl)-5H-pyrano[3,2-c]benzopyran-5-ones 9a and 9b, respectively (Scheme 3). The formation of compound 9a,b could be visualised according the simple Michael addition of the carbanion (C_3 -coumarinyl) to the β -carbon of the activated nitrile followed by 1,6-exo-dig cyclisation and transesterification. Structure 9a gets more chemical evidence through the reaction with phenacyl chloride, furoyl chloride and hydrazine hydrate. Thus, treatment of compound 9a with phenacyl chloride in refluxing pyridine yielded 9benzoyl-7,10-dihydro-8-hydroxy-7-(4-methoxyphenyl)-4H-[1]benzopyrano[3',4':5,6]pyrano[2,3-b]pyrrol-6-one (10). Similarly, acylation of compound 9a with furoyl chloride in boiling pyridine afforded methyl-2-furoylamino-4-(4methoxyphenyl)-5-oxo-5H-pyrano[3,2-c]benzopyran-3carboxylate (11). Hydrazinolysis of compound 9a using hydrazine hydrate (80%) afforded a crude solid product (two spots in TLC) which upon purification yielded salicylic acid hydrazide (12) and 3-(5-amino-3-oxopyrazolin-4-yl)-3-(4methoxyphenyl) propanoic acid hydrazide (13).

Experimental

Melting points were taken on a Griffin and George melting point apparatus. IR spectra were recorded on a Pye Unicam SP 1200



Reagents: a, PhCOCH2CI/Py; b, 2-furoyl chloride/Py; c, N2H4.H2O/EtOH

spectrophotometer using the KBr wafer technique. ¹H NMR spectra were determined on a Varian Gemini 200 MHz using TMS as internal standard. ¹³C NMR spectra were measured on a JEOL 75 MHz spectrometer. EI MS were measured on a Shimadzu-GC-MS, QP 1000 EX instrument operating at 70 eV. Elemental analyses were carried out at the Microanalytical Unit, Faculty of Science, Ain Shams University with a Perkin-Elmer Series II CHNS/O elemental analyser. The homogeneity of the synthesised compounds was checked using TLC with aluminium sheets silica gel F₂₅₄ (Merck).

10-Acetyl-7,9-dihydroxy-6H-benzo[c]chromen-6-one (2): Ethyl coumarin-3-carboxylate (4.4 g, 0.02 mol), dry sodium ethoxide (1.1 g, 0.02 mole) and pentane-2,4-dione (2.4 ml, 0.024 mole) were stirred together in a flask heated in an oil-bath at 170 °C for 3 h. After cooling, the reaction mixture was treated with cold dilute hydrochloric acid. The solid that separated was filtered off, washed several times with water, dried, and recrystallised from ethanol as orange crystals (67%), m.p. 192–194 °C. IR: v_{max} 3449 br (OH), 1697 (lactone CO), 1673 cm⁻¹ (chelated ketone CO). NMR (CDCl₃): $\delta_{\rm H}$ 8.1–7.1 (m, 4H), 6.7 (s, 1H, C8-H), 5.3 (br.s, 2H, 2OH), 2.4 (s, 3H, CH₃); $\delta_{\rm C}$, see inset structure. EI MS: m/z (%) 270 (M⁺, 51), 255 (M–CH₃, 100), 227 (M–COCH₃, 96), 171 (8.5), 143 (12), 115 (19), 89 (20), 77 (42). Anal. Calcd for C1₅H₁₀O₅ (270.24): C, 66.66; H, 3.72. Found: C, 67.0; H, 3.49%.



¹³C NMR of compound **2**

3-Aryl-2,3-dihydrochromeno[6,5-c]chromene-1,7-diones (3a-d): Compound 2 (2.7 g, 0.01 mol), an aromatic aldehyde (0.01 mol) and piperidine (0.5 ml) were refluxed in n-butanol (50 ml) for 6 h, when no more substrate was detected (TLC). The solid deposited from the hot solution was filtered off, dried and recrystallised to give 3a-d.

3-(4-Methoxyphenyl) compound (**3a**): Recrystallised from AcOH as yellow crystals (73%), m.p. 230–232 °C. IR: v_{max} br. 3500 (OH), 1687, 1670 cm⁻¹ (CO). NMR (DMSO-4₆): δ_{H} 7.7–6.5 (m, 9H_{ar}), 5.8 (s, 1H, exchangeable with D₂O), 5.3 (t, H_a, J_{ac} = 5.8 Hz, J_{ab} = 6.4 Hz), 3.9 (s, 3H, OMe), 3.2 (d, d, H_c, J_{cb} = 16.8 Hz, J_{ac} = 7.1 Hz), 2.9 (d, d, H_b, J_{bc} = 17.4 Hz, J_{ab} = 5.7 Hz). ¹³C NMR (DMSO-4₆); δ_{C} 190.2 (CO), 50.6 (C2), 48.7 (C3), 107.3 (C5), 166.4 (C6) 108.2 [(C6a), 159.3 (CO), 153.3 (C–O), 123.7, 128.3, 126.4, 127.3, 130.7, 148.6 coumarinyl carbons], 164.1 (C4a), 123.7, 127.8, 118.2, 158.3 C_{ar}, 55.4 (MeO). EI MS: m/z (%) 388 (M⁺, 100), 255 (29), 227 (3.4), 134 (73), 120 (60). Anal. Calcd for C₂₃H₁₆O₆ (388.37): C, 71.13; H, 4.15. Found: C, 71.41; H, 4.08%.

3-(4-Dimethylaminophenyl) compound (**3b**): Recrystallised from dioxan as red crystals (65%), m.p. 277–279 °C. IR: v_{max} 3473 br (OH), 1692, 1678 cm⁻¹ (CO). NMR (DMSO-d₆): $\delta_{\rm H}$ 7.9–6.6 (m, 9H_{ar}), 5.9 (s, 1H, OH, exchangeable with D₂O). 5.1 (t, H_a, J_{ac} = 6.1 Hz, J_{ab} = 6.7 Hz), 3.4 (d,d, H_c, J_{bc} = 17.1 Hz, J_{ac} = 6.7 Hz), 3.1 (d,d, H_b, J_{bc} = 17.6 Hz, J_{ab} = 6.2 Hz), 2.85 (s, 6H, NMe₂). EI MS: *m*/z (%) 401 (M⁺, 35), 357 (63), 120 (100). Anal. Calcd for C₂₄H₁₉NO₅ 401.42): C, 71.81; H, 4.77; N, 3.48. Found: C, 72.01; H, 4.51; N, 3.77%.

3-(4-Chlorophenyl) compound (3c): Recrystallised from AcOH as orange crystals (51%), m.p. 280–281 °C. IR: v_{max} 3472 br (OH), 1687, 1670 cm⁻¹ (CO). NMR (DMSO-d₀: δ_{H} 8.1–7.2 (m, 9H_{ar}), 5.6 (s, 1H, exchangeable with D₂O), 4.99 (t, H_a, J_{ac} = 5.2 Hz, J_{ab} = 7.1 Hz), 3.3 (d,d, H_c, J_{bc} = 15.8 Hz, J_{ac} = 6.44 Hz), 3.0 (d,d, H_b, J_{bc} = 16.8 Hz, J_{ba} = 6.0 Hz). EI MS: m/z (%) 394 (M⁺, 32.4), 392 (M⁺, 100). Anal. Calcd for C₂₂H₁₃ClO₅ (392.79): C, 67.27; H, 3.33; Cl, 9.02. Found: C, 67.51; H, 3.50; Cl, 9.20%.

3-(β-*Styryl) compound* (**3d**): Recrystallised from AcOH as orange crystals (60%), m.p. 268–269°C. IR: v_{max} 3492 br (OH), 1688, 1669 cm⁻¹ (CO). EI MS: *m/z* (%) 384 (M⁺, 33), 281 (66), 104 (100), 77 (41). Anal. Calcd for C₂₄H₁₆O₅ (384.39): C, 74.99; H, 4.19. Found: C, 75.28; H, 4.37%.

Reaction of **2** with ethyl cyanoacetate; formation of 6-hydroxy-1methyl-3,7-dioxo-3H,7H-chromeno[4,3-f]chromene-2-carbonitrile (**4**): A mixture of the acetyl compound **2** (2.7 g, 0.01 mole) and ethyl cyanoacetate (1.13 g, 0.01 mole) was stirred in the presence of sodium metal (0.5 g, 0.02 mole) in dry xylene on an oil-bath at 180° C for 3 h. The reaction mixture was cooled and poured into cold dilute hydrochloric acid. The deposited solid was filtered off, washed several times with water, dried and then recrystallised from acetic acid to give **4** as light brown crystals (34%), m.p. 340–342 °C. IR: v_{max} 3437 br (OH), 2228 (C=N), 1736 (lactone CO), 1671 cm⁻¹ (chelated ketone CO). NMR (DMSO-d₆): $\delta_{\rm H}$ 7.6–6.6 (m, 5H_{ar}), 5.2 (s, 1H, OH), 1.67 (s, 3H, CH₃); $\delta_{\rm C}$ 128.1, 125.9, 127.7, 124.3 (C9–C12), 152.3 (C8a), 131.4 (C12a), 161.1 (C7), 108.6 (C6a), 160.9 (C6), 109.2 (C5), 157.7 (C3), 96.4 (C2), 115.3 (C=N), 170.1 (C1), 14.4 (CH₃), 118.4, 144.6, 130.9. EI MS: m/z (%) 319 (M⁺, 84), 291(M-CO, 100), 262 (33), 76 (40). Anal. Calcd for C₁₈H₉NO₅ (319.27): C, 67.71; H, 2.84; N, 4.38. Found: C, 68.08; H, 2.60; N, 4.6%.

Reaction of **2** with hydrazine: formation of azine **5**: To a solution of **2** (2.7 g, 0.01 mole) in n-butanol, hydrazine hydrate (80%) (0.01 mole) was added with stirring. The mixture was refluxed for 8 h until no more substrate remained (TLC). Evaporation of the excess solvent left a solid product which recrystallised from acetic acid as orange crystals (35%), m.p. 346–348 °C. IR: v_{max} 3413 br (OH), 1693 (CO), 1634 cm⁻¹ (C=N). NMR (DMSO-d₆): $\delta_{\rm H}$ 7.8–6.67 (m, 10H_{ar}), 5.1 (br. s, 4H), 2.1 (s, 6H). EI MS: *m/z* (%) 536 (M⁺, 28), 268 (100), 228 (60). Anal. Calcd for C₃₀H₂₀N₂O₈ (536.50): C, 67.16; H, 3.75; N, 5.22. Found: C, 66.87; H, 3.52; N, 5.45%.

Reaction of **2** with phenyl hydrazine; formation of 5-hydroxy-1methyl-3-phenylchromeno[4,3-e]indazol-6(3H)-one (**6**): A mixture of compound **2** (2.7 g, 0.01 mole) and phenylhydrazine (1.08 g, 0.01 mole) was heated on an oil-bath at 180 °C for 3 h (TLC). The reaction mixture was then poured into ice-cold hydrochloric acid. The solid obtained was filtered off, washed several times with water, dried and recrystallised from acetic acid to give the chromeno-indazole derivative **6** as yellow crystals (24%), m.p. 265–267 °C. IR: v_{max} 3422, 1691 (CO), 1640 cm⁻¹ (C=N). NMR (DMSO-d₆): $\delta_{\rm H}$ 7.8–6.9 (m, 10H_{ar}), 5.3 (s, 1H), 2.6 (s, 3H). EI MS: m/z (%) 342 (M⁺, 100), 282 (36), 238 (14), 77 (66). Anal. Calcd for C₂₁H₁₄N₂O₃ (342.33): C, 73.68; H, 4.11; N, 8.18. Found: C, 73.72; H, 4.16; N, 8.44%.

Reaction of **2** with primary amines; formation of 10-[1-(substituted imino)ethyl]-7,9-dihydroxy-6H-benzo[c]chromen-6-one **7a–h**: To a solution of **2** (0.9 g, 0.0033 mole) in n-butanol, a primary amine (p-toluidine, benzylamine, aniline, ethylamine, n-butylamine, dodecylamine, hexdecylamine or octdecylamine) was added and the whole mixture was refluxed for 6 hrs until no more substrate (TLC) was detected. The reaction mixture was poured onto ice-cold hydrochloric acid. The solid deposited was washed several times with water, dried and recrystallised from the indicated solvent to give **7a–h**.

p-Tolylimine **7a**: Yellow crystals (78%) from AcOH, m.p. 214–216 °C. IR: v_{max} 3462 br. (OH), 1680 (CO), 1640 cm⁻¹ (C=N). NMR (DMSO-d₆): δ_{H} 8–7.2 (m, 8H_{ar}), 6.3 (s, 1H, C₈-H), 4.66 (br.s, 2H, OH exchangeable with D₂O), 2.3 (s, 3H, Ar-Me), 0.9 (s, 3H, Me). EI MS: *m/z* (%) 359 (M⁺, 100), 344 (58), 254 (44), 91 (41), 65 (38). Anal. Calcd for C₂₂H₁₇NO₄ (359.36): C, 73.53; H, 4.76; N, 3.89. Found: C, 73.74; H, 4.61; N, 3.67%.

 $\begin{array}{l} \textit{Benzylimine 7b: pale yellow crystals (71\%) from benzene, m.p. 145-146 ^{\circ}C. IR: v_{max} 3436 br (OH), 1682 (CO), 1632 cm^{-1} (C=N). NMR (DMSO-d_6): δ_{H} 7.8-7.1 (m, 9H_{ar}), 6.48 (s, 1H, C8-H), 5.0 (br.s, 2H, exchangeable with D_2O), 4.81 (s, 2H, CH_2Ph), 1.2 (s, 3H, Me). EI MS: m/z (%) 359 (M⁺, 33), 263 (100), 91 (82), 65 (32). Anal. Calcd for C_{22}H_{17}NO_4(359.36): C, 73.53; H, 4.76; N, 3.89. Found: C, 73.74; H, 4.61; N, 3.67\%. \end{array}$

Phenylimine **7c**: Recrystallised from toluene as pale yellow crystals (67%), m.p. 150–152 °C. IR: v_{max} 3426 br (OH), 1676 (CO), 1628 cm⁻¹ (C=N). NMR (CDCl₃): $\delta_{\rm H}$ 7.7–7.1 (m, 9H_{ar}), 6.4 (s, 1H, C₈–H), 4.7 (br.s, 2H, exchangeable with D₂O), 1.08 (s, 3H, Me). EI MS: *m/z* (%) 345 (M⁺, 100). Anal. Calcd for C₂₁H₁₅NO₄ (345.33): C, 73.04; H, 4.37; N, 4.05. Found: C, 72.91; H, 4.17; N, 4.20%.

Ethylimine **7d**: Recrystallised from AcOH as pale yellow crystals (61%), m.p. 207–209 °C. IR: v_{max} 3470 br (OH), 1686 (CO), 1632 cm⁻¹ (C=N). NMR (DMSO-d₆): $\delta_{\rm H}$ 7.6–7.1 (m, 4H_{ar}), 6.3 (s, 1H, C₈-H), 3.56 (q, 2H), 1.2 (t, 3H), 0.99 (s, 3H, Me); $\delta_{\rm C}$ see inset structure. Anal. Calcd for C₁₇H₁₅NO₄ (297.29): C, 68.68; H, 5.07; N, 4.71. Found: C, 68.76; H, 5.2; N, 4.52%.



¹³C NMR of compound 7d

¹³C NMR of compounds 7e

n-Butylimine 7e: Recrystallised from toluene as pale yellow crystals (56%), m.p. 165–167°C. IR: ν_{max} br. 3444 (OH), 1688 (CO), 1638 cm²l (C=N). NMR (CDCl_3): δ_H 7.8–7.2 (m, 4H_ar), 6.6 (s, 1H, C_8-H), 5.2 (br. s, 2H, exchangeable with D₂O), 3.6 (t, 2H), 1.7-1.3 (m, 4H), 1.08 (s, 3H, Me), 0.92 (s, 3H, Me); δ_C , see inset structure. EI MS: m/z (%) 325 (M⁺, 71), 268 (100), 226 (43). Anal. Calcd for C₁₉H₁₉NO₄ (325.34): C, 70.14; H, 5.87; N, 4.30. Found: C, 70.32; H, 5.60; N, 4.55%

n-Dodecylimine 7f: Recrystallised from benzene as pale yellow crystals (52%), m.p. 142–143 °C. IR: ν_{max} 3462 br (OH), 1682 (CO), 1633 cm⁻¹ (C=N). NMR (CDCl₃): $\delta_{\rm H}$ 7.6–7.1 (m, 4H_{ar}), 6.4 (s, 1H, C₈-H), 5.1 (br.s, 2H, exchangeable with D₂O), 3.6 (t, 2H), 1.66-1.3 (m, 20H), 1.03 (t, 3H), 0.94 (s, 3H, Me). EI MS: m/z (%) 437 (M⁺, 66), 268 (100). Anal. Calcd for $C_{27}H_{35}NO_4$ (437.53): C, 74.12; H, 8.05; N, 3.20. Found: C, 74.42; H, 7.76; N, 3.0%.

n-Hexadecylimine 7g: Recrystallised from light petroleum ether as pale yellow crystals (49%), m.p.110–111 °C. IR: ν_{max} 3446 br (OH), 1688 (CO), 1636 cm⁻¹ (C=N). NMR (CDCl₃): $\delta_{\rm H}$ 7.8–7.2 (m, 4H_{ar}), 6.6 (s, 1H, C₈-H), 5.19 (br.s, 2H), 3.56 (t, 2H), 1.69–1.36 (m, 28H), 1.03 (t, 3H), 0.99 (s, 3H, Me). EI MS: m/z (%) 493 (M⁺, 40), 267 (100). Anal. Calcd for C31H43NO4 (493.63): C, 75.43; H, 8.76; N, 2.83. Found: C, 75.29; H, 8.49; N, 2.61%.

n-Octadecylimine 7h: Recrystallised from benzene as pale yellow crystals (56%), m.135–137 °C. IR: ν_{max} 3470 br (OH), 1686 (CO), 1636 cm⁻¹ (C=N). NMR (CDCl₃): $\delta_{\rm H}$ 7.7–7.2 (m, 4H_{ar}), 6.48 (s, 1H, C_8 -H), 4.98 (br.s, 2H), 3.54 (t, 2H), 1.65–1.3 (m, 32H), 1.1 (t, 3H), 0.97 (s, 3H, Me). EI MS: m/z (%) 521 (M+, 17), 268 (100). Anal. Calcd for C₃₃H₄₇NO₄ (521.68): C, 75.97; H, 9.06; N, 2.68. Found: C, 75.77; H, 9.24; N, 2.91%.

6-Hydroxy-1-methyl-3-thioxo-3,4-dihydrochromeno[4,3-f] quinazolin-7-one (8): A mixture of 2 (2.7 g, 0.01 mole) and thiourea (1.52 g, 0.02 mole) was dissolved in acetic acid and heated under reflux for 3 h (TLC). The reaction mixture was poured onto cold water and the solid that separated was filtered off, washed with dilute ethanol, dried and recrystallised from dioxan to give 8 as deep yellow crystals (41%), m.p. 183-185 °C. IR: v_{max} 3492 br (NH, OH), 1691 (CO), 1632 (C=N), 1324 cm⁻¹ (C=S). NMR (DMSO-d₆): $\delta_{\rm H}$ 7.3–6.6 (m, 5H_{ar}), 5.3 (br.s, 1H), 3.0 (br.s, 1H), 1.5 (s, 3H, Me). EI MS: m/z (%) 310 (M⁺, 11), 282 (100), 223 (60), 76 (32). Anal. Calcd for C₁₆H₁₀N₂O₃S (310.25): C, 61.94; H, 3.24; N, 9.03; S, 10.32. Found: C, 62.34; H, 2.98; N, 8.82; S, 10.07%.

Formation of the pyrano-chromenones 9a and 9b: A mixture of 4hydroxycoumarin (1.62 g, 0.01 mole), sodium methoxide (0.5 g sodium in 50 ml absolute methanol) and ethyl-a-cyano-4-methoxycinnamate or a-cyano-4-methoxycinnamonitrile (0.01 mole) was refluxed with stirring for 3 h (TLC). The reaction mixture was poured on cold dilute acetic acid and allowed to stand for 1 h at room temperature. The solid that separated in each case was filtered off, washed with water, dried and recrystallised from the indicated solvent.

2-amino-4-(4-methoxyphenyl)-5-oxo-4,5-dihydropyrano Methvl [3,2-c]chromene-3-carboxylate (9a): Yellow crystals (66%) from benzene, m.p. 165-167 °C. IR: v_{max} 3422, 3306, 3271 (NH₂), 1695 (CO unsaturated δ-lactone), 1658 cm⁻¹ (chelated CO ester). NMR (CDCl₃): δ_H 7.4–6.9 (m, 8H_{ar}), 4.7 (s, 1H, C₄-H), 4.0 (br.s, 2H, exchangeable with D₂O), 3.85 (s, 3H, OMe), 3.7 (s, 3H, COOMe); $\delta_{\rm C}$, see inset structure. EI MS: m/z (%) 379 (M⁺, 38), 320 (35), 272 (99), 249 (69), 240 (100), 145 (27), 121 (64), 117 (52), 89 (64). Anal. Calcd for C₂₁H₁₇NO₆ (379.35): C, 66.49; H, 4.51; N, 3.69. Found: C, 66.70; H, 4.21; N, 3.82%.

2-Amino-4-(4-methoxyphenyl)-5-oxo-4,5-dihydropyrano[3,2-c] chromene-3-carbonitrile (9b): Yellow crystals (52%) from acetic acid, m.p. 231–233 °C. IR: ν_{max} 3399, 3322, 3171 (NH_2), 2194 (C=N), 1709 cm⁻¹ (CO unsaturated δ -lactone). NMR (DMSO-d₆): $\delta_{\rm H}$ 7.3–6.9 (m, 8H_{ar}), 4.7 (s, 1H, C₄-H), 4.0 (br.s, 2H, exchangeable with D₂O), 3.85 (s, 3H, OMe); $\delta_{\rm C}$, see inset structure. EI MS: m/z (%) 346 (M⁺, 17), 280 (92), 279 (100), 249 (75), 240 (26), 145 (22), 121 (19), (89 (40), 66 (74). Anal. Calcd for $C_{20}H_{14}N_2O_4$







¹³C NMR of compound **9b**

(346.32): C, 69.36; H, 4.06; N, 8.09. Found: C, 69.52; H, 4.11; N. 8.43%

3-Aryl-2,3-dihydrochromeno[6,5-c]chromene-1,7-dionIf 9-Benzoyl-8-hydroxy-7-(4-methoxyphenyl)-7,10-dihydro-6H-chromeno[3',4':5,6] pyrano[2,3-b]pyrrol-6-one (10)

Phenacyl chloride (1.5 g, 0.01 mole) was added dropwise with stirring to the ester 9a (3.79 g, 0.01 mole) in pyridine (20 ml) over 1 h and the mixture was then heated under reflux for another 1 h, monitoring the progress of the reaction by TLC. Evaporation of the solvent in vacuo left a semisolid product which was triturated with acetone and the solid which separated was filtered off, dried and recrystallised from dioxan to give compound 10 as yellow crystals (41%), m.p. 324–326°C. IR: v_{max} 3425 br (NH, OH), 1727 (CO), 1661 cm⁻¹ (chelated ketone CO). NMR (DMSO-d₆): $\delta_{\rm H}$ 8.3–7.2 (m, 14H_{ar} + NH), 6.8 (br.s, 1H), 4.7 (s, 1H, C₄-H), 3.65 (s, 3H, OMe). EI MS: m/z (%) 447 (M⁺-H₂O, 15.9), 424 (56.0), 317 (100), 92 (13.0). Anal. Calcd for C₂₈H₁₉NO₆ (465.44): C, 72.25; H, 4.11; N, 3.01. Found: C, 72.49; H, 4.19; N, 3.0%.

Methvl 2-furoylamino-4-(4-methoxyphenyl)-5-oxo-4,5-dihydropyrano[3,2-c]chromene-3-carboxylate (11): Furoyl chloride (2 g, 0.015 mol) was added dropwise with stirring to a solution of the ester 9a (3.79 g, 0.01 mole) in pyridine (20 ml), and the mixture was refluxed for 2 h (TLC). The reaction mixture then poured onto crushed ice containing acetic acid. The solid that separated was filtered off, dried and recrystallised from light petroleum ether (b.p. 80-100°C) to give 11 as yellow crystals (63%), m.p. 102-104°C. IR: v_{max} 3270 (NH), 1730 (CO unsaturated δ -lactone), 1710 (CO ester), 1670 cm⁻¹ (CO amide). NMR (CDCl₃): $\delta_{\rm H}$ 8.7 (s, 1H, NH), 8.18–6.82 (m, 11H_{ar}), 4.8 (s, 1H, C₄-H), 3.91 (s, 3H, OMe). EI MS: *m/z* (%) 473 (M⁺, 10), 280 (77), 279 (82), 250 (64), 121(26), 92 (100), 64 (48). Anal. Calcd for C₂₆H₁₉NO₈ (473.41): C, 65.96; H, 4.04; N, 2.95. Found: C, 66.10; H, 4.0; N, 3.2%.

Hydrazinolysis of ester 9a; formation of 12 and 13: A solution of 9a with hydrazine hydrate (80%) in molar ratio 1:3 in absolute ethanol (30 ml) was heated under reflux for 8 h (TLC). Evaporation of the excess solvent left a crude solid product (two spots in TLC) which triturated with diethyl ether. The residue was filtered off, dried and recrystallised from ethanol to give salicylic acid hydrazide 12. Evaporation of ether left a solid product which recrystallised from acetic acid to give 13.

3-(5-Amino-3-oxopyrazolin-4-yl)-3(4-methoxyphenyl)propanoyl hydrazide (13):Yellowish-white crystals (41%), m.p. 156-158°C. IR: v_{max} 3462, 3376, 3290, 3176 (NH₂), 1680 cm⁻¹ (CO). NMR (CDCl₃): $\delta_{\rm H}$ 8.3 (s, 2H, 2NH, exchangeable with D₂O), 7.1–6.7 (m, 4H_{ar}), 4.9



¹³C NMR of compound 13

(t, H_a, J_{ac} 5.99 Hz, J_{ab} 6.8 Hz), 3.9 (s, 3H, OMe), 3.3 (d,d, H_c, J_{cb} 16.8 Hz, J_{ac} 7.1 Hz), 3.1 (d,d, H_b, J_{bc} 17.2 Hz, J_{ba} 6.6 Hz), 2.2 (br.s, 5H, exchangeable with D₂O). EI MS: m/z (%) 263 (M⁺-N₂, 100), 260 (67), 232 (17), 185 (22), 77 (80). Anal. Calacd for C₁₃H₁₇N₅O₃ (291.29): C, 53.6; H, 5.87; N, 24.04. Found: C, 53.92; H, 5.41; N, 23.74%.

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References

- 1 I. Manolov and N.D. Danchev; J. Med. Chem. Chim. Ther., 1995, 30, 531.
- 2 O. Egan, R. O'Kennedy, E. Moran, D. Cox, E. Prosser and R.D. Thornes, *Drug. Metab. Rev.*, 1990, **22**, 503.
- A.M. El-Sayed and O.A. Abd-allah; *P, S, Si, Relat. Elem.*, 2001, **170**, 75.
 B. Kalluraya, P. Vishwanatha, A.M. Isloor, G. Rai and M. Kotian; *Boll. Chim. Farm.*, 2000, **139**, 263.
- 5 O.A. Abd-allah; Farmaco, 2000, 55, 641.
- 6 A.M. Al-Agrody, M.S. Abd-El-Latif, N.A. El-Hady, A.H. Fekry and A.H. Bedair; *Molecules*, 2001, 6, 519.
- 7 A.A. Emmanuel-Giota, K.C. Fylaktakidou, D.J. Hadjipavlou-Litina, K.E. Litinas and D.N. Nicolaides; J. Heterocyclic Chem., 2001, 38, 717.
- Z.N. Noval, M.I. El-Zahar and S.S. Abd El-Karim; *Molecules*, 2000, 5, 99.
 L. Raev, E. Voinov, I. Ivanov and D. Popov, *Pharmazie*, 1990, 45, 696;
- *Chem. Abstr.* 1990, **114**, 74 711b.

- F.M. Dean, Naturally occurring oxygen ring compounds, Butterworths, London, 1963.
- A. Mustafa, Furopyrans and furopyrones, Wiley Interscience, New York, 1967.
- 12 F.F. Abdel-Latif, Indian J. Chem., 1990, 29B, 664.
- 13 F.F. Abdel-Latif, M.M. Mashaly, R. Mekheimer and T.B. Abdel-Aleem, Z. Naturforsch., 1993, 48b, 817.
- 14 F.F. Abdel-Latif and R.M. Shaker, Bull. Soc. Chim. Fr, 1991, 127, 87.
- 15 R.M. Shaker, Pharmazie, 1996, 51, 148.
- 16 M.R. Mahmoud, E.A.A. El-Bordany, N.F. Hassan and F.S.M. Abu El-Azm, J. Chem. Res., 2007, 541.
- 17 M.R. Mahmoud, E.A.A. El-Bordany, N.F. Hassan and F.S.M. Abu El-Azm, J. P,S,Si and Relat. Elem., 2007, 181, 1275.
- 18 M.R. Mahmoud, H.M.F. Madkour, E.A.A. El-Bordany and E.A. Soliman, J. Chem. Res., 2007, 673.
- 19 M.R. Mahmoud, M.M. El-Shahawi and S.E. Farahat, J. Chem. Res., 2008, 86.
- 20 M.R. Mahmoud, M.M. El-Shahawi and S.E. Farahat, J. Chem. Res., 2008, 59.
- 21 A. Sammour, M. Abdalla and M. El-Kady, *Acta Chim. Acad. Sci. Hung.*, 1974, 82, 369.
- 22 D.H. Williams and I. Fleming, Spectroscopic methods in organic chemistry, Mc Graw-Hill, 3rd edn, 1980.
- 23 D. Heber, Arch. Pharm., 1987, 320, 577.
- 24 D. Heber, I.C. Ivanov and S.K. Karagiosov, J. Heterocyclic Chem., 1995, 32, 505.